Addition of Fluorene to Schiff Bases[†]

Kostadinka Popandova-Yambolieva

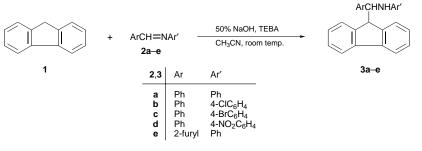
Department of Chemistry, University of Sofia, 1 J.Bourchier Avenue, 1126 Sofia, Bulgaria

Addition of fluorene to N-arylmethylideneanilines under conditions of phase-transfer catalysis gives N-aryl-N-[9H-fluoren-9-yl(aryl)methyl]amines 3a-e.

It is known that indene and fluorene have acidic properties, but their application in the Michael addition has been little described. The addition of fluorene to some α,β -unsaturated ketones,¹ nitriles and esters² has been realized under various conditions. Some years ago we reported our preliminary results on the Michael addition of fluorene to some α,β -unsaturated ketones, esters and nitriles under conditions of phase-transfer catalysis (PTC) in acetonitrile.3 Later a similar addition was realized with other α,β -unsaturated ketones and nitriles under PTC conditions in benzene. There is no report in the literature on the addition of fluorene to Schiff bases. Having in mind that some fluorenyl derivatives of aminobenzoic acids are anti-inflammatory agents, we decided to study the behaviour of fluorene (1) towards *N*-arylmethylideneanilines (2). We now present the results of our studies on the reaction of **1** with **2** under PTC conditions. The expected compounds **3a–e** were obtained (Scheme 1).

The reaction was carried out at room temperature with an excess of aqueous sodium hydroxide, a catalytic amount of

mmol) in acetonitrile (10 cm³). The reaction mixture was stirred at room temperature for 1 h. Water (100 cm³) was added and the solid was filtered off, washed until neutral and recrystallized from methanol-ethyl acetate. The following products were obtained: N-[9H-fluoren-9-yl(phenyl)methyl]-N-phenylamine (3a) (1.48 g, 43%), mp 152–153 °C; v_{max} /cm⁻¹ (CHCl₃) 3430 (NH) (Found: C, 89.78; H, 6.3; N, 3.9. C₂₆H₂₁N requires C, 89.9; H, 6.05; N, 4.0%); δ_{H} 3.18 (1 H, br s, NH), 4.48 (1 H, d, J 4.5 Hz), 5.18 (1 H, d, J 4.5 Hz), 6.15-7.80 (18 H, m, aromatic); N-(4-chlorophenyl)-N-[9H-fluoren- $^{-1.50}$ (CDCl₃) $^{-1.50$ (17 H, aromatic); N-(4-*bromophenyl*)-N-[9H-*fluoren*-9-yl(*phenyl*-)*methyl*]-*amine* (**3c**) (0.7 g, 17%), mp 170–172 °C; v_{max}/cm^{-1} (CHCl₃) 3420 (NH) (Found: C, 73.1; H, 4.7; N, 3.4, C₂₆H₂₀BrN requires C, 73.2; H, 4.7; N, 3.3%); $\delta_{\rm H}$ 3.36 (1 H, br s, NMH), 4.40 (1 H, d, J 4.0 Hz), 5.11 (1 H, d, J 4.0 Hz), 6.10–7.80 (17 H, m, aromatic); N-[9H-*fluoren*-9-yl(*phenyl*)*methyl*]-N-(4-*nitrophenyl*)*a*-*mine* (**3d**) (0.8 g, 20%, mp 127–128 °C; v_{max}/cm⁻¹ (CHCl₃) 3370



Scheme 1

TEBA (triethylbenzylammonium chloride) and acetonitrile as solvent. For 4-methoxy- and 4-dimethylamino-benzylide-neanilines only the starting Schiff bases were recovered. Also, owing to steric hindrance, there was no reaction with either N-phenylbenzophenone imine or with N-benzylidenetert-butylamine.

The structures of the new compounds 3a-e were assigned by IR and ¹H NMR spectroscopy and by elemental analysis.

Experimental

Melting points were determined by a Boetius micropoint apparatus and are uncorrected. IR spectra were recorded on a Specord 71 spectrophotometer (Carl Zeiss, Iena). ¹H NMR spectra were measured with a TESLA BS 487-C spectrometer (80 MHz) using CDCl₃ solutions and Me₄Si as internal standard. The starting Narylmethylideneanilines were prepared by reported procedures.

Typical Procedure. - Aqueous sodium hydroxide (50% w/v; 3 cm³) was added to a stirred solution of fluorene (1.66 g, 10 mmol), the corresponding Schiff base 2 (10 mmol) and TEBA (0.23 g, 1

(NH), 1590 (NO₂) (Found: C, 79.3; H, 5.1; N, 6.9. $C_{26}H_{20}N_2O_2$ (NH), 1590 (NO₂) (Found: C, 79.3; H, 5.1; N, 6.9. $C_{26}H_{20}N_2O_2$ requires C, 79.5; H, 5.1; N, 7.1%); N-[9H-*fluoren-9-yl*(2-*fjurl*)*me thyl*]-N-*phenylamine* (**3e**) (0.95 g, 29%), mp 178–180 °C; v_{max}/cm^{-1} (CDCl₃) 3430 (NH) (Found: C, 85.5; H, 5.7; N, 3.9. $C_{24}H_{19}NO$ requires C, 85.4; H, 5.7; N, 4.1%). δ_H 3.46 (1 H, br s, NH), 4.68 (1 H, d, *J* 4.1 Hz), 5.26 (1 H, d, *J* 4.1 Hz), 6.04 and 6.30 (2 H, 2 d, *J* 6; *J* 6 Hz), 6.62 (1 H, d, *J* 14 Hz), 6.90–7.75 (13 H, m, aromatic).

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^{*}To receive any correspondence.

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